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Attorney File Ref: 102790-122 / 30058

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

Application of:

Anjum Fatima SHAUKAT

Serial No .:

10/524699

Filed:

16.February 2005

Examiner:

Dr. John HARDEE

Art Group:

1751

PARFOMAK

Title:

FRAGRANCED BLEACHING COMPOSITION

PER TELEFAX: 571 273-8300

Mail Stop: Amendment Commissioner for Patents P.O. Box 1450 Alexandria, VA 22313 - 1450

Dear Sir;

AFFIDAVIT OF CO-INVENTOR

- I, ANJUM FATIMA SHAUKAT, British citizen, of 7, Crane Road, Twickenham, Middlesex TW2 6RX, England, do solemnly declare and attest as follows:
- I am a co-inventor of an invention entitled "Fragranced Bleaching Composition", 1. which is the subject of an application for Letters Patent in the United States of America, which was filed on February 16, 2005 under the Serial Number 10/524,699, and which was published in the United States as US 2005/0227892 A1.

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- 2. I hold the degrees of Graduateship of the Royal Society of Chemistry, Part 1, awarded in 1987 by University of Huddersfield and Graduateship of the Royal Society of Chemistry, Part 2, awarded in 1989 by Kingston University.
- 3. Since 1999 I have been employed on a part-time consultancy basis by Givaudan (U.K.) Limited, a British company, of Magna House, 76-80, Church Street. Staines, Middlesex TW18 4XR, England, at the Givaudan Centre for Innovative Technology, in which capacity I am called upon to participate in the originating and development of innovative technologies.
- 4. I have performed the experiments as detailed in the following:

I have repeated Example B1 of Table 1 on p.19 of PCT published application WO 02/44314 of Henkel KGaA. I also have prepared an example according to the abovementioned 10/524,699, specifically Example A of Table 1, as shown on p.7 of US 2005/0227892 A1.

The compositions are shown in the following table:

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Formulation:	E 1	Example A
Raw material	%wt.	%wt.
Deionised water	q.s. to 100	q.s. to 100
Sodium hypochlorite (12% active)	0.00	14.10
C10-13 alkyl benzenesulphonic acid	4.00	0.00
C12-22 fatty alcohol + 7EO	0.00	0.00
C12-15 alcohol ethoxylate, 3EO	0.00	4.44
C12-18 fatty alcohol + 7EO*	2.00	0.00
Citric acid 1H ₂ 0	6.50	0.00
Amine oxide (30% active)	0.00	9.92
Sulfosuccinate	0.00	1.78
Sulphate	0.00	1.47
Oleic acid	0.00	1.78
Isopropanol	0.00	5.33
Potassium Hydroxide	5.90	0.00
Sodium hydroxide (50% active)	0.00	1.31
Isoparaffinic hydrocarbon	0.00	11.11
Dioctyl Ether	4.50	0.00
Hydrogen Peroxide	0.45	0.00
Perfume	1.50	0.20
Dye	< 0.01	0.00
Glutaraldebyde	0.05	0.00

- * The material C12-22 fatty alcohol + 7EO, used in the original E1, was not available, and it was replaced by the readily-available and functionally equivalent C12-18 fatty alcohol + 7EO (DehydrolTM LT7 ex Cognis GmbH, named as a preferred compound (bevorzugte Vertreter) in WO 02/44314 at p.12, 1.2-4).
- 5. The two mixtures were agitated until completely homogeneous and then allowed to stand.
- 6. The example from 10/524,699 (Example A) separated into two clear separate layers with no turbidity in under 6 hours. On the other hand, the Example E1 took over 11 hours to achieve the same state.

I hereby declare that all statements made herein of my own knowledge are true and that all statements made on information and belief are believed to be true, and further that

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these statements were made with the knowledge that wilful false statements and the like, so made, are punishable by fine or imprisonment or both, under Section 1001 of Title 18 of the United States Code, and that such wilful false statements may jeopardize the validity of the patent application or any patent issued thereon

Respectfully submitted:

Anium Fatima SHAUKA'I

25.9.07

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